

## School of Engineering Mid-Term Examinations - November 2024

Semester: 5 <sup>th</sup>			Date: 04.11.2024					
Course Code: PET2031Time			Time:	09:30am – 11:00am				
Course Name: Overview of Material Science Max M			Max M	<b>1arks</b> : 50				
Pro	Program: B. Tech Weigh							
Instructions: (i) Read all questions carefully and answer accordingly. (ii) Do not write anything on the question paper other than roll number. Part A								
Ans	swer Al	LL the Questions. Each question carries 2 marks.		5Q x	2M=1	0M		
1		e the term "Material".	2 Marks	L1		C01		
2	List two points on the importance of material science in the <b>2 Mark</b> petroleum refinery industry.			L1	C01			
3	List fo	our characteristics of a polymeric material.	2 Marks	L1 C01				
4	Reproduce the coordination number of a unit cell with a2 Markssuitable schematic diagram.			L1	CO2			
5	Defin	e Schottky defects with a suitable image.	2 Marks	L1 CO2		CO2		
Part B								
Answer ALL Questions. Each question carries 10 marks.					4QX10M=40M			
	6a	Define composite material with a suitable example.		2 Marks	L1	CO1		
6	6b	<b>6b</b> Explain the difference between metal and ceramic material.		3 Marks	L2	C01		
	<b>6c</b> Suppose you are a leader or expert in "Material Science" division in a reputed industry. Being a leader, explain elaborately any <b>five</b> important parameters for selecting a right material from the thousands available.			5 Marks	L3	<b>CO1</b>		

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	7a	Define opaque body with an example.	2 Marks	L1	CO1
7	7b	<b>All ceramic materials are not crystalline</b> - explain this with proper mechanism and example.		L2	CO1
	7c	Imagine you are tasked with designing a material for a smartphone screen that needs to be transparent, mechanical strength, and high thermal resistance. Choose the best-fitted or suitable material type from the followings: crystalline, semi- crystalline, or amorphous material. Also explain the reasons behind for not selecting the other types material very clearly.	5 Marks	L2	<b>C01</b>
8	8a	Discuss face-centered cubic (FCC) crystal structure with suitable diagram.	2 Marks	L2	CO1
	8b	Discuss the number of atoms (N) per unit cell of an FCC crystal.	3 Marks	L2	CO1
	8c	Demonstrate the unit cell volume and atomic packing factor (APF) of an FCC crystal. Also solve for the percent of void space present in that particular type of crystal structure. (Consider all spherical atoms are having same radius, which is "R")	5 Marks	L3	C01
		or			
9	9a	Discuss simple cubic (SC) crystal structure with suitable diagram.	2 Marks	L2	CO1
	9b	Discuss the number of atoms (N) per unit cell of an SC crystal.	3 Marks	L2	C01
	9c	Demonstrate the unit cell volume and atomic packing factor (APF) of an SC crystal. Also solve for the percent of void space present in that particular type of crystal structure. (Consider all spherical atoms are having same radius, which is "R")	5 Marks	L3	<b>C01</b>
10	10a	Discuss Hock's Law with a suitable diagram.	2 Marks	L2	CO2
	10b	Compare between ductile and brittle material, three points only.	3 Marks	L2	CO2
	10c	Explain how stress and strain influence the failure of a metal under tensile loading.	5 Marks	L2	CO2

	11a	Define fatigue failure.	2 Marks	L1	CO2
11	11b	Discuss Brinell hardness and Rockwell hardness, and their applications.	3 Marks	L2	CO2
	11c	An automotive engine (consider a car) was tested at a fixed working conditions as follows:	5 Marks	L3	CO2
		Load: 30 N/m²; Temperature: Ambient (25°C); Car Speed: 80 km/hr.			
		Suddenly, the working load of the engine is increased to 55 N/m <sup>2</sup> and the car is suddenly stroked with the wall. After that it is seen that the metal body of the car and engine was gone down permanently. Discuss your own design aspect in terms of selection of a material, safety of car so that the car can get a minimum loss after a striking with the same speed, i.e., 80 km/hr.			
12	12a	Discuss cup and cone type fracture of a solid material with proper diagram.	2 Marks	L2	CO2
	12b	Discuss clearly the effect of temperature and thermal shock on the fracture of a material taking a suitable example.	3 Marks	L2	CO2
	12c	Illustrate clearly the stress-strain curve for the following type of polymeric materials. From the graph select the material that has more sustainability at high temperature.	5 Marks	L3	CO2
		<ul><li>(a) Elastomer</li><li>(b) Strain-softening (quasi brittle material)</li><li>(c) Ductile</li><li>(d) Brittle</li></ul>			
or					
	13a	Define the term "Ultimate Safety Factor (USF)".	2 Marks	L1	CO2
13	13b	In the manufacturing of critical components for a bridge, the material chosen must exhibit consistent mechanical properties across all parts. However, due to inherent property variability in materials, different batches may exhibit slight differences in strength, ductility, or toughness. Explain property variability impact the reliability and safety of the structure	8 Marks	L2	CO2

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