



# PRESIDENCY UNIVERSITY

BENGALURU

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## Mid - Term Examinations - MARCH 2026

Date: 11-03- 2026

Time: 09:30am - 11:00am

<b>School:</b> SOCSE	<b>Program:</b> B Tech	
<b>Course Code :</b> PHY2501	<b>Course Name:</b> Optoelectronics and Quantum Physics	
<b>Semester:</b> II	<b>Max Marks:</b> 50	<b>Weightage:</b> 25%

CO - Levels	CO1	CO2	CO3	CO4	CO5	CO6
Marks	26	24				40

### Instructions:

- (i) Read all questions carefully and answer accordingly.
- (ii) Do not write anything on the question paper other than roll number.
- (iii) Given : Planck's constant  $h = 6.625 \times 10^{-34}$  Js; Boltzmann's constant  $k_B = 1.38 \times 10^{-23}$  J/K, Speed of light  $c = 3 \times 10^8$  m/s, Mass of the electron  $m = 9.1 \times 10^{-31}$  kg, Charge of the electron  $e = 1.6 \times 10^{-19}$  C

## Part A

Answer ALL the Questions. Each question carries 2marks.

5Q x 2M=10M

1	An electronic circuit requires a device that maintains constant voltage even when input voltage fluctuates, Identify the suitable type of the semiconductor.	2 Marks	L2	CO1
2	Determine the wavelength of light emitted from LED which is made up of GaAsP semiconductor whose forbidden energy gap is 1.875 eV.	2 Marks	L3	CO1
3	Describe superconductivity and discuss the importance of critical temperature.	2 Marks	L2	CO1
4	Explain Schrodinger's time independent wave equation and state its physical meaning.	2 Marks	L2	CO2
5	Illustrate the identity matrix and present its representation for a $2 \times 2$ system.	2 Marks	L3	CO2

## Part B

### Answer the Questions.

**Total Marks 40M**

<b>6.</b>	<b>a.</b>	<p>A home is planning to install a solar panel system to reduce electricity cost. Apply your knowledge of solar cells to explain how the solar cell will generate electricity. Use the principle, construction, working, and efficiency concepts in your explanation.</p>	<b>10 Marks</b>	<b>L3</b>	<b>CO1</b>
	<b>b.</b>	<p>Differentiate intrinsic and extrinsic semiconductors.</p> <p>In a P-type germanium, <math>n_i = 2.1 \times 10^{19} \text{ m}^{-3}</math> density of boron <math>4.5 \times 10^{23} \text{ atoms /m}^3</math>. The electron and hole mobility are <math>0.4 \text{ m}^2 \text{ v}^{-1} \text{ s}^{-1}</math> and <math>0.2 \text{ m}^2 \text{ v}^{-1} \text{ s}^{-1}</math> respectively. Find its conductivity before and after addition of boron atoms.</p>	<b>10 Marks</b>	<b>L3</b>	<b>CO6</b>
<b>Or</b>					
<b>7.</b>	<b>a.</b>	<p>Apply the principle of Hall effect to determine whether a given current carrying semiconductor placed in a perpendicular magnetic field is n-type or p-type, and use appropriate diagrams and mathematical expressions for Hall voltage and Hall coefficient to support your explanation.</p> <p>Calculate the Hall voltage when a conductor carrying a current of 100 A, is placed in a magnetic field of 1.5 T. The conductor has a thickness of 1 cm, and the number density of charges inside the conductor is <math>5.9 \times 10^{28} /\text{m}^3</math>.</p>	<b>10 Marks</b>	<b>L3</b>	<b>CO1</b>
	<b>b.</b>	<p>A paramagnetic material has a magnetic field intensity of <math>10^4 \text{ Am}^{-1}</math>. If the susceptibility of the material at room temperature is <math>3.7 \times 10^{-5}</math>. Calculate the magnetization and flux density in the material.</p> <p>Apply your understanding of superconducting materials to compare and distinguish between soft superconductors and hard superconductors based on their properties and applications.</p>	<b>10 Marks</b>	<b>L3</b>	<b>CO6</b>
<b>8.</b>	<b>a.</b>	<p>Apply the concept of Heisenberg's Uncertainty Principle to explain its physical significance with suitable examples, describe the important properties of a wave function.</p> <p>What is the minimum uncertainty in the energy state of an atom if an electron remains in this state for <math>10^{-8} \text{ s}</math>.</p>	<b>10 Marks</b>	<b>L2</b>	<b>CO2</b>
	<b>b.</b>	<p>An electron in a molecule travel at a speed of 40 m/s. The uncertainty in the momentum <math>\Delta p</math> of the electron is <math>10^{-6}</math> of its</p>	<b>10 Marks</b>	<b>L3</b>	<b>CO6</b>

		<p>momentum. Compute the uncertainty in position <math>\Delta x</math> if the mass of an electron is <math>9.1 \times 10^{-31}</math> kg.</p> <p>Mention the properties of matter waves</p>			
<b>Or</b>					
<b>9.</b>	<b>a.</b>	Apply your knowledge of Pauli matrices to demonstrate their properties and use suitable examples to illustrate their operations.	<b>10 Marks</b>	<b>L3</b>	<b>CO2</b>
	<b>b.</b>	<p>Let <math> u_1\rangle,  u_2\rangle,  u_3\rangle</math> be an orthonormal basis. In this basis, define</p> <p><math> \alpha\rangle = i u_1\rangle + 2 u_2\rangle - (1+i) u_3\rangle</math>  <math> \beta\rangle = 3 u_1\rangle - i u_2\rangle + 2 u_3\rangle</math></p> <p>(a) Find <math>\langle\alpha </math> and <math>\langle\beta </math>.  (b) Compute <math>\langle\alpha \beta\rangle</math> and verify that <math>\langle\alpha \beta\rangle = \langle\beta \alpha\rangle^*</math>.  (c) If <math>b = 1 + 2i</math>, compute <math> b\alpha\rangle</math>.  (d) Find <math> \alpha + \beta\rangle</math> and <math> \alpha - \beta\rangle</math>.</p>	<b>10 Marks</b>	<b>L3</b>	<b>CO6</b>