



PRESIDENCY UNIVERSITY

BENGALURU

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End - Term Examinations – May / June 2026

Date: 20- 05-2026

Time:01:00pm – 04:00pm

School: SOE	Program: B. Tech-PET	
Course Code: PET2023	Course Name: Design in Production Engineering	
Semester: VI	Max Marks: 100	Weightage: 50%

CO - Levels	C01	C02	C03	C04	C05	C06
Marks	5	5	5	29	28	28

Instructions:

- (i) Read all questions carefully and answer accordingly.
- (ii) Do not write anything on the question paper other than roll number.

Part A

Answer ALL the Questions. Each question carries 2marks.

10Q x 2M=20M

1.	Identify the main components of a heater treater.	2 Marks	L1	C04
2.	Recognize two common emulsion-treating mechanisms inside a heater treater.	2 Marks	L1	C04
3.	Compute the heat input when oil is heated from 20°C to 90°C.	2 Marks	L2	C05
4.	Differentiate between latent heat and sensible heat.	2 Marks	L2	C05
5.	Describe how density difference is related to settling velocity using Stokes' law.	2 Marks	L2	C05
6.	Predict what will happen to settling velocity if droplet size is very small.	2 Marks	L2	C05
7.	Classify the types of heat exchangers based on construction.	2 Marks	L2	C06
8.	Define any two parts of a shell and tube heat exchanger.	2 Marks	L1	C06
9.	Describe the fouling phenomenon in heat exchangers.	2 Marks	L2	C06
10.	Describe the law governing conduction mode of heat transfer.	2 Marks	L2	C06

Part B

Answer the Questions.

Total Marks 80M

11.	a.	(i) Describe how charge size and formation properties affect the performance of perforation in oil wells, with a suitable diagram. (ii) Describe the configuration of a gas lift system considering different valve arrangements and production flow conditions.	5+5 Marks	L2	CO1, CO2
	b.	(i) A vertical well of depth 2800 m has a tubing head pressure of 850 kPa and casing head pressure of 3900 kPa. The tubing gradient is 11 kPa/m and gas gradient is 4.5 kPa/m. If 4 unloading valves are installed, Using the constant pressure drop method, determine the depth at which each valve should be placed. (ii) Determine the operating conditions that control the separation of emulsions in a heater treater.	5+5 Marks	L3	CO3, CO4
Or					
12.	a.	(i) Classify the perforation methods used in well completion and mention the advantages of jet perforation. (ii) Explain the surface and subsurface equipments of Gas lift.	5+5 Marks	L2	CO1, CO2
	b.	(i) A marginal well is to be operated using intermittent gas lift based on the fallback approach. Given data: Tubing inside diameter = 2.25 in, Average liquid fallback velocity = 0.7 m/s, Injection gas pressure = 5200 kPa, Tubing pressure gradient = 12 kPa/m, Valve opening pressure differential = 400 kPa. Apply the fallback method to determine the depth of the unloading valve. (ii) Determine the different methods employed for emulsion separation in heater treaters and explain their basic principles.	5+5 Marks	L3	CO3, CO4
13.	a.	Determine the equation governing particle settling in a fluid using Stoke's law and explain its application in crude oil demulsification.	10 Marks	L3	CO4
	b.	Determine the key variables that control gravity separation processes and explain the influence of temperature on phase separation performance.	10 Marks	L3	CO4
Or					
14.	a.	Examine the important parameters influencing the performance and sizing of gravity-based separation systems in petroleum processing.	10 Marks	L3	CO4
	b.	Determine how a heater treater facilitates phase separation by gravity and illustrate the process with a clear diagram.	10 Marks	L3	CO4
15.		Determine the capacity of a heater treater vessel required to achieve proper gravity separation under the following conditions: Droplet size = 120 μm , Viscosity (μ) = 0.025 Pa·s, Density difference ($\Delta\rho$) = 180 kg/m ³ , Height = 2.0 m, Flow rate = 220 m ³ /day. Further, what will happen to the settling velocity when the droplet size is reduced by 50% from its initial value.	20 Marks	L3	CO5
Or					

16.	5 Kg of ice sample initially at -5°C is heated to steam at 120°C use the data given below to determine the total heat required. $C_{\text{ice}} = 2 \text{ KJ/Kg. }^{\circ}\text{C}$, $h_f = 300 \text{ KJ/Kg}$, $C_{\text{water}} = 4.18 \text{ KJ/Kg. }^{\circ}\text{C}$, $h_{fg} = 2000 \text{ KJ/Kg}$, $C_{\text{stem}} = 2 \text{ KJ/Kg}$. Show each stage of heat addition clearly. Further, distinguish between latent heat of vaporization and specific heat capacity.	20 Marks	L3	C05
17.	Breakdown the methodology followed in the thermal design of a shell-and-tube heat exchanger and analyze the impact of factors like heat duty estimation, log mean temperature difference, fouling considerations, and heat transfer coefficient on the overall design in surface production operations.	20 Marks	L3	C06
Or				
18.	Examine how changes in hydrocarbon characteristics like viscosity, heat capacity, and density affect heat exchanger performance, and evaluate their implications on exchanger dimensions, number of tube passes, and overall heat transfer requirements.	20 Marks	L3	C06