



PRESIDENCY UNIVERSITY

BENGALURU

Roll No.												
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End - Term Examinations – May / June 2026

Date: 25-05-2026

Time: 01:00pm – 04:00pm

School: SOCSE	Program: B. Tech.	
Course Code: CHE3003	Course Name: Introduction to Nanotechnology and Applications	
Semester: VI	Max Marks: 100	Weightage: 50%

CO - Levels	C01	C02	C03	C04	C05
Marks	14	12	14	24	36

Instructions:

- (i) Read all questions carefully and answer accordingly.
- (ii) Do not write anything on the question paper other than roll number.

Part A

Answer ALL the Questions. Each question carries 2marks.

10Q x 2M=20M

1.	Define nanotechnology.	2 Marks	L1	C01
2.	Define core-shell nanoparticles.	2 Marks	L1	C02
3.	What are ceramic nanomaterials?	2 Marks	L2	C03
4.	List two examples of metal nanoparticles.	2 Marks	L1	C03
5.	State how the band energy of a nanoparticle changes with size.	2 Marks	L2	C04
6.	Define surface plasmon resonance (SPR).	2 Marks	L1	C05
7.	Name nanomaterials used in sunscreen.	2 Marks	L1	C05
8.	Differentiate between passive and active RFID (any 2 points)	2 Marks	L2	C05
9.	Define magic numbers in nanomaterials.	2 Marks	L1	C04
10.	Explain top-down approach used in nanomaterial synthesis.	2 Marks	L2	C01

Part B

Answer the Questions.

Total Marks 80M

11.	a.	Differentiate between bulk materials and nanomaterials.	5 Marks	L2	C01
	b.	Explain the size-dependent properties of nanomaterial.	5 Marks	L2	C01
Or					
12.	a.	Discuss in detail the bottom-up approaches for the synthesis of nanomaterials with suitable examples.	5 Marks	L2	C01
	b.	Calculate the surface-to-volume ratio of a spherical nanomaterial of radius 33 nm.	5 Marks	L3	C01

13.	a.	Describe the classification of nanomaterials as 0D, 1D, 2D, and 3D, with examples.	6 Marks	L2	C02
	b.	Determine the emission wavelength of quantum dots having an effective band gap of 3.5 eV.	4 Marks	L3	C02
Or					
14.	a.	Calculate the quantum confinement energy change when the radius of a quantum dot is decreases from 20 nm to 5 nm.	5 Marks	L3	C02
	b.	Discuss the types of nanocomposites in details.	5 Marks	L2	C02

15.	a.	Define carbon nanotubes (CNTs), mention their chirality, and distinguish between single-walled and multi-walled carbon nanotubes.	10 Marks	L2	C03
Or					
16.	a.	Define metal oxide nanoparticles, give examples, and explain their properties, applications, and challenges.	10 Marks	L2	C03

17.	a.	Discuss the optical properties of nanomaterials. Explain how size influences optical behavior and relate it to blue and red shift.	10 Marks	L2	C04
Or					
18.	a.	Explain the concept of quantum confinement and analyze its effect on the energy levels of nanomaterials.	10 Marks	L2	C04

19.	a.	Calculate the number of atoms in tetrahedral close-packed nanoclusters for n=3, 6, and 9 using the appropriate formula. Compare the results and determine which cluster is more stable.	10 Marks	L4	C04
Or					

20.	a.	Analyze the relative stability of 147, 150, and 309 atom gold nanoparticles based on closed-shell configurations.	10 Marks	L4	C04
21.	a.	Explain Radio Frequency Identification (RFID) and discuss its working principle with the help of a neat diagram.	10 Marks	L2	C05
Or					
22.	a.	Explain how nanotechnology is used in solar cells and how it improves their efficiency.	10 Marks	L2	C05
23.	a.	Describe the construction and the working principle of electrochemical glucose sensors, including the relevant chemical reactions.	10 Marks	L2	C05
Or					
24.	a.	Explain the construction and working principle of an electrochemical methanol (VOC) sensor with chemical reactions.	10 Marks	L2	C05
25.	a.	Discuss the advantages of using nanomaterials in drug delivery and how drugs are loaded onto nanomaterials.	10 Marks	L2	C05
Or					
26.	a.	Explain superabsorbent nanomaterials. Mention the materials used, their properties, and applications in health and hygiene.	10 Marks	L2	C05