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Max Marks: 100



# PRESIDENCY UNIVERSITY BENGALURU

# SCHOOL OF ENGINEERING END TERM EXAMINATION - DEC 2022

Semester: Semester III - 2021 Date: 13-JAN-2023

**Course Code**: PET2009 **Time**: 1.00PM - 4.00PM

Course Name: Sem III - PET2009 - Thermodynamics of Reservoir Fluids

**Program :** B.Tech. Petroleum Engineering **Weightage :** 50%

## Instructions:

(i) Read all questions carefully and answer accordingly.

(ii) Question paper consists of 3 parts.

(iii) Scientific and non-programmable calculator are permitted.

### **PART A**

10 X 2 = 20M
(OO4) [[(a and a dual
(CO1) [Knowledge]
(CO2) [Knowledge]
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(CO3) [Knowledge]
(CO3) [Knowledge] [Page 1 of 2]

#### **PART B**

#### ANSWER ALL THE FOUR QUESTIONS

 $4 \times 10 = 40M$ 

**11.** "The efficiency of heat engine is always less than unity". Give a mathematical proof of this statement and relate it with second law of thermodynamics.

(CO1) [Comprehension]

**12.** "A certain property of the system remains constant during phase transition", Identify the property and mathematically prove the above statement.

(CO1) [Comprehension]

**13.** With the help of neat and labeled diagram, state the common classifications of crude oils and state the salient features of these crude oil.

(CO3) [Comprehension]

**14.** A certain process follows the relationship PV = constant. Identify the process and obtain the relationship for work, heat and internal energy.

(CO1) [Comprehension]

#### **PART C**

#### ANSWER ALL THE TWO QUESTIONS

 $2 \times 20 = 40M$ 

- 15. (a) "A turbine is an essential device used in power plants which assists in generation of electricity. It contains elements that rotate. This rotation is transferred to generator which generates the electricity". In one such turbine steam enters at 350°C and 2 MPa. The exit from the turbine is a saturated vapor at 50 kPa. Determine the efficiency of the turbine (*Use the provided Mollier Diagram and attach with answer script*).
  - (b) Describe the process termed as PT flash. List the places where a crude oil might experience PT flash.

(CO2) [Application]

- **16.** (a) A solution contains 78.0 gram of glucose (MM = 180.16 g/mol) in 500 grams of water at 25°C. The vapor pressure of pure water at this temperature is 23.8 mm Hg. Based on the data given above:
  - (i) Identify and state the theory/law which you can would apply to determine the vapor pressure of the solution.
  - (ii) Based on the identified theory/law estimate the vapor pressure of the solution based on the data given above. (Assume MM of water to be 18.02 g/mol)
  - (b) Calculate Degree of freedom for :
    - (i) Liquid water in equilibrium with its vapor.
    - (ii) Liquid water in equilibrium with a mixture of water vapor and nitrogen.
    - (iii) A three-phase system of a saturated aqueous salt solution at its boiling point with excess salt crystals present.

(CO1) [Application]

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