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**PRESIDENCY UNIVERSITY
BENGALURU**

**SCHOOL OF ENGINEERING
MID TERM EXAMINATION - APR 2023**

Semester : Semester II - 2022

Course Code : PHY1001

Course Name : Sem II - PHY1001 - Material Physics

Program : CIV

Date : 13-APR-2023

Time : 2:00PM - 3:30PM

Max Marks : 50

Weightage : 25%

Instructions:

- (i) Read all questions carefully and answer accordingly.
- (ii) Question paper consists of 3 parts.
- (iii) Scientific and non-programmable calculator are permitted.
- (iv) Do not write any information on the question paper other than Roll Number.

PART A

ANSWER ALL THE QUESTIONS

(5 X 2 = 10M)

1. What is the use of X-ray diffraction?
(CO1) [Knowledge]
2. What are primitive unit cells and what are non-primitive unit cells?
(CO1) [Knowledge]
3. Explain the terms (i) Basis (ii) Space lattice (iii) Unit cell
(CO1) [Knowledge]
4. What is the formula to calculate the strain?
(CO2) [Knowledge]
5. Write few properties of ceramic materials?
(CO2) [Knowledge]

PART B

ANSWER ALL THE QUESTIONS

(4 X 5 = 20M)

6. Deduce packing factors for body-centered cubic (BCC) structure.
(CO1) [Comprehension]
7. Derive the relationship between atomic radius and lattice parameter for SCC and FCC.
(CO1) [Comprehension]
8. A brass wire of length 2 m has one end, fixed to a rigid support, and from the other end, a 4 kg wt is suspended. If the cross-sectional area of the wire is $12.25 \times 10^{-8} m^2$. Find the extension produced in the wire (Young's modulus of wire = $11 \times 10^{10} N/M^2$).
(CO2) [Comprehension]

9. A bar 500 mm long and 22 mm in diameter is elongated by 1.2 mm under the effect of axial pull of 105 KN. Calculate stress, strain and modulus of elasticity (Young's Modulus).

(CO2) [Comprehension]

PART C

ANSWER ALL THE QUESTIONS

(2 X 10 = 20M)

10. (a) Describe the cubic, tetragonal, and hexagonal crystal systems. (6 Marks)
(b) What are one-dimensional defects. Explain in a few words. (4 Marks)

(CO1) [Application]

11. Chromium has a BCC structure. its atomic radius is 0.1 nm. Calculate the free volume of the unit cell.

(CO1) [Application]